Theoret. Chim. Acta (Berl.) 45, 309–315 (1977)

Topological Properties of Benzenoid Systems

An Identity for the Sextet Polynomial

Ivan Gutman

Institute "Rudjer Bošković", P.O.B. 1016, 41001 Zagreb, Croatia, Yugoslavia

The sextet polynomial of non-branched cata-condensed benzenoid molecules is proved to be related (Eq. (3)) to the characteristic polynomial of a tree.

Key words: Benzenoid systems, topological properties of \sim

1. Introduction

Several recent investigations [1, 2] have shown that resonance theory gives satisfactory quantitative results when applied to benzenoid hydrocarbons. The combinatorial background of this theory was also studied [3, 4]. Hosoya and Yamaguchi demonstrated [4] that various resonance-theoretical approaches can be unified by means of the sextet polynomial. For a benzenoid hydrocarbon G, the sextet polynomial $B_G(X)$ is defined as

$$B_{\rm G}(X) = \sum_{k=0}^m r({\rm G}, k) X^k$$

where r(G, k) is the so called "resonant sextet number", namely the number of ways in which k mutually resonant sextets can be selected from G. m is the maximal number of mutually resonant sextets. By definition, r(G, 0)=1. Further details on these resonance-theoretical notions can be found in Refs. [4, 5].

Let the benzenoid molecule possess R rings. Then instead of the sextet polynomial we can consider another polynomial $BB_G(X)$,

$$BB_{\rm G}(X) = X^{R+1}B_{\rm G}(-1/X^2) = \sum_{k=0}^{\left[(R+1)/2\right]} (-1)^k r({\rm G}, k) X^{R+1-2k}$$
(1)

The form of $BB_G(X)$ resembles much to the characteristic polynomial of a tree (i.e. of an acyclic graph) with R+1 vertices (see Eq. (2)).

No simple and generally valid algorithm for the calculation of the sextet polynomials is known at present. The usual way of determining $B_G(X)$ by considering all possible selections of resonant sextets in G is not only rather tedious, but is also unreliable for complicated polycyclic systems. In the present paper it will be shown that the calculation of $B_G(X)$ for non-branched cata-condensed benzenoid (NBCCB)¹ molecules is reduced to the finding of the characteristic polynomial of a tree *T*. This tree can be easily deduced from the structure of the NBCCB network.

Before starting with the consideration of the properties of the sextet polynomial, we shall describe a class of trees.

Let P_n denote the path² with *n* vertices. The tree $T(t_1, t_2, ..., t_n)$ is obtained by joining t_j new vertices to the *j*'th vertex of P_n (j=1, 2, ..., n). Thus t_j is an integer or zero. If $t_1 = t_2 = \cdots = t_n = 0$, then $T(t_1, t_2, ..., t_n) \equiv P_n$. The number of vertices of $T(t_1, t_2, ..., t_n)$ is $N = t_1 + t_2 + \cdots + t_n + n$. For example we present T(4), T(2, 1, 0, 0, 0) and T(1, 1, 3, 0, 0, 2).



The characteristic polynomial of a tree T has the form [7]:

$$P_T(X) = \sum_{k=0}^{\lfloor N/2 \rfloor} (-1)^k p(T, k) X^{N-2k}$$
(2)

where p(T, k) is the number of ways in which k non-incident edges can be selected from T.

The analogy between expressions (2) and (1) is evident. In the following we will demonstrate that there exists indeed a tree T such that for a particular class of benzenoid molecules,

$$BB_{G}(X) = P_{T}(X) \tag{3}$$

¹ A molecular network which is composed entirely of regular hexagons is called benzenoid. If no three hexagons have a common atom, the system is called cata-condensed (in the opposite case, the system is peri-condensed). If every hexagon of a cata-condensed system has at most two neighbouring hexagons, it is said to be non-branched. [6]

² The "path" (or "chain") with *n* vertices is the graph with the properties that the *j*'th vertex is adjacent to the (j-1)'th and (j+1)'th, and only to them (j=2, 3, ..., n-1).

2. Some Topological and Resonance-Theoretical Properties of NBCCB Molecules

Let us consider an *R*-cyclic NBCCB system G and denote by R(j) (j=1, ..., R) its rings. These rings will be numbered successively, so that the ring R(j) is neighbouring to the rings R(j-1) and R(j+1). Hence, R(1) and R(R) are terminal rings. Of course [4, 5], R(j) presents also a resonant sextet.

A ring in a NBCCB hydrocarbon can be annelated exactly in two distinct ways, namely in a linear (L) or in an angular (A) mode.



Therefore, the ring R(j) can be labelled by a symbol S_j , where either $S_j = L$ or $S_j = A$. The two terminal rings can have arbitrary labels, but it is convenient to denote them by L. Thus an ordered sequence $S = S(G) = S_1 S_2 \dots S_R$ of L and A symbols (an "L, A-sequence") can be associated with every NBCCB system G. For example, molecules 1 and 2 have S(1) = LLLL and S(2) = LAALAALAL.



For brevity we will denote LL by L^2 , LLL by L^3 etc.

In a general case the L, A-sequence S is of the form

$$S = (L)^{s_1} A(L)^{s_2} A \dots (L)^{s_{n-1}} A(L)^{s_n}$$
(4)

where s_j are non-negative integers. In other words, the rings $R(s_1+1)$, $R(s_1+s_2+2)$, ..., $R(s_1+s_2+\cdots+s_{n-1}+n-1)$ are assumed to be of A-mode, while all others of the L-mode. Since G has R rings, $R=s_1+s_2+\cdots+s_n+n-1$.

Theorem 1. Eq. (3) holds for every G which is a NBCCB hydrocarbon. If the L, A-sequence of G is given by Eq. (4), then $T = T(s_1, s_2, ..., s_n)$.

Proof. If T is a tree with R edges (and therefore with R + 1 vertices), such that for all i, j the edges e_i, e_j are mutually non-incident if R(i) and R(j) are mutually resonant, then

p(T, k) = r(G, k)

from which Theorem 1 is deduced straightforwardly. Therefore, in order to demonstrate the validity of Theorem 1, it is sufficient to prove the following

Theorem 2. The edges of the graph $T(s_1, s_2, ..., s_n)$ can be labelled in such a way that any pair of edges e_i, e_j is non-incident, whenever R(i) and R(j) are mutually resonant in G, with S(G) being given by Eq. (4).

Proof. Let us consider the ring R(p) of G. Then the sequence S is of the form

$$S = \dots A(L)^{a-1} S_p(L)^{b-1} A \dots$$
(5)

where a and b are integers. In other words, we assume that R(p-a) and R(p+b) are the nearest (relative to R(p)) angularly annelated rings.

Now, it can be verified straightforwardly that the sextets $R(p-a), \ldots, R(p-1)$, $R(p+1), \ldots, R(p+b)$ can never be resonant if the sextet R(p) is resonant. This occurs simply because the double bonds in $R(p-a), \ldots, R(p-1), R(p+1), \ldots$, R(p+b) become localized after R(p) is deleted from G. Let us illustrate this on two examples. The resonant sextet R(p) is indicated by a circle.



 $S_p = L, a = 2, b = 3$



 $S_p = A, a = 2, b = 3$

Furthermore, since the double bonds in the rings $R(1), \ldots, R(p-a-1), R(p+b+1), \ldots, R(R)$ are not localized when R(p) is deleted from G, the sextets R(p), R(i) and R(j) are mutually resonant, where $i=1, \ldots, p-a-1$ and $j=p+b+1, \ldots, R$.

This discussion leads to the following conclusion.

- a) If $S_p = A$, any two sextets R(i), R(j) are mutually resonant for i=p-a+1,..., p-1 and j=p+1,...,p+b-1.
- b) If $S_{v} = L$, no two such sextets R(i), R(j) are mutually resonant.

Let us now represent every ring R(j) of G by an edge e_i and try to construct a tree T with the properties required in Theorem 2. Then statements a) and b) read,

- a*) If $S_p = A$, any two edges e_i , e_j are mutually non-incident if i = p a + 1, ..., p 1 and j = p + 1, ..., p + b 1.
- b*) If $S_p = L$, any two of such edges e_i , e_j are incident.

This situation can be uniquely realized by the graph fragments 3 (for $S_p = A$) and 4 (for $S_p = L$): a^{**}) If $S_p = A$



$$b^{**}$$
) If $S_n = L$



But the graph fragments 3 and 4 are exactly Theorem 2 applied to Eq. (5). The extension of a^{**}) and b^{**}) to the whole sequence S given by Eq. (4), completes finally the proof of Theorem 2.

Corrolary 1. Every tree $T(t_1, t_2, ..., t_n)$ can be associated with an L, A-sequence (Eq. (4)) and thus with a class of NBCCB molecules.

For example, the trees T(4), T(2, 1, 0, 0, 0) and T(1, 1, 3, 0, 0, 2) correspond to sequences LLLL, LLALAAA = LLALAAL and LALALLLAAALL, respectively.

Corrolary 2. If two NBCCB systems have the same L, A-sequence, and therefore the same associated tree, they are indistinguishable from the viewpoint of resonance theory. The smallest such molecules are 5 and 6; both have an LAAL sequence, with P_5 being their associated tree.

Using Theorem 1, it is rather simple to construct the tree T associated with any NBCCB hydrocarbon. For example, the trees corresponding to molecules 1 and 2 are T(4) and T(1, 0, 1, 0, 1, 1), respectively.



Finally, it is important to note that the validity of Theorem 1 cannot be simply extended, neither to branched cata-condensed, nor to peri-condensed molecules. Namely, for the smallest branched cata-condensed (7) and peri-condensed (8) benzenoid systems, the corresponding r(G, k) values cannot be related to the coefficients of the characteristic polynomial of any graph.



3. Discussion

Theorem 1 is, as to the author's knowledge, the first case that graph spectral theory plays some role in the theory of conjugated compounds *independently of the Hückel molecular orbital model*. The main applicability of our result is in determining $B_{\rm G}(X)$'s. There are well known simple graph-theoretical techniques [8] for the calculation of the characteristic polynomial of a tree. Hence, for example, the characteristic polynomials of T(4), T(1, 0, 1, 0, 1, 1) and T(1, 0, 1) are $X^5 - 4X^3$, $X^{10} - 9X^8 + 26X^6 - 27X^4 + 8X^2$ and $X^5 - 4X^3 + 3X$, respectively. Therefore, $B_1(X) = 1 + 4X$, $B_2(X) = 1 + 9X + 26X^2 + 27X^3 + 8X^4$ and $B_5(X) = B_6(X) = 1 + 4X + 3X^2$. The evaluation of $B_2(X)$ without the use of our theorem would be rather tedious.

Since $B_G(1)$ is equal to the number of Kekulé structures K = K(G) of the molecule G [4], from Eq. (3) it follows,

$$K(G) = \sum_{k=0}^{\lfloor (R+1)/2 \rfloor} p(T,k) = Z_T$$

where Z_T is the topological index of Hosoya [7, 9]. The K value of the molecule **2** (=71) could hardly be obtained by "brute force" enumeration.

As a final application we mention that recently Aihara [2] found that the roots of

Topological Properties of Benzenoid Systems

 $B_G(X)$ can be used for the calculation of the resonance energy (and thus aromaticity) of G. Because of Eq. (3), these roots are closely related to the spectrum of T, and therefore the resonance energy of a NBCCB hydrocarbon is obtained from the spectrum of its associated tree T.

References

- Herndon, W. C.: J. Am. Chem. Soc. 95, 2404 (1973); J. Org. Chem. 40, 3583 (1975); Herndon, W. C., Ellzey, M. L.: J. Am. Chem. Soc. 96, 6631 (1974); Swinborne-Sheldrake, R., Herndon, W. C., Gutman, I.: Tetrahedron Letters, 755 (1975); Aihara, J.: Bull. Chem. Soc. Japan 49, 1429 (1976); Randić, M.: Chem. Phys. Letters 38, 68 (1976)
- 2. Aihara, J.: private communication
- 3. Gutman, I., Herndon, W. C.: Chem. Phys. Letters 34, 387 (1975)
- 4. Hosoya, H., Yamaguchi, T.: Tetrahedron Letters, 4659 (1975)
- 5. Clar, E.: The aromatic sextet. London: Wiley 1972
- Balaban, A. T., Harary, F.: Tetrahedron 24, 2505 (1968); Balaban, A. T.: Tetrahedron 25, 2949 (1969); Gutman, I.: Croat. Chem. Acta 46, 209 (1974)
- 7. Hosoya, H.: Theoret. Chim. Acta (Berl.) 25, 215 (1972)
- Heilbronner, E.: Helv. Chim. Acta 36, 170 (1953); Harary, F., King, C., Mowshowitz, A., Read, R. C.: Bull. London Math. Soc. 3, 321 (1971)
- 9. Hosoya, H.: Bull. Chem. Soc. Japan 44, 2332 (1971)

Received March 1, 1977